# Interatomic Potentials and the Dipole Moment of the Cadmium Forbidden $(5^{3}P_{2} \rightarrow 5^{1}S_{0})$ Transition for the Cd<sub>2</sub>System

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The molecular band appearing at  $\Delta V = 770 \text{ cm}^{-1}$  on the blue wing of the intercombination Cd (326.1 nm) line broadened by Cd pressure has been investigated. This band was interpreted as a rainbow satellite accompanying the dipoleforbidden Cd (314.1 nm) line associated with the  $5^{3}P_{2} \rightarrow 5^{1}S_{0}$  transition. The profile of this band has been calculated using the semiclassical model developed by Szudy and Baylis. To obtain this profile the interatomic potential of the molecular state  $1_{u}$  and the dipole moment associated with this band have been modified to give the best fit profile with the experimental one.

Key words: Interatomic potentials, Dipole moment, Cadmium.

## Introduction

The Cd, molecule is representative of a class of excimer molecules. For the Cd dimer, quantitative experimental information on interatomic potentials available only for the lowest excited states (Kowalski et al 1985; Helmi et al 1996; Czajkowski et al 1989 and Bousquet 1986). Moreover, the existing theoretical potentials (Bender et al 1979 and Stevens 1979) for Cd, may be taken only as a rough guide to the spectroscopy of this system. In a previous paper (Helmi et al 1996), we noticed that there are three bands (A, B and C) at about 770, 830 and 1800 cm<sup>-1</sup> respectively have been observed experimentally on the blue wing of the Cd (326.1 nm) line. The interpretation of these bands have explained in (Grycuk et al 1992). The band A was interpreted as a rainbow satellite accompanying the dipole - forbidden Cd line 314.1 nm associated with the  $5^{3}P_{2} \rightarrow 5^{1}S_{0}$  transition where the potential difference between the upper and lower states has an extremum. As this band has been observed at 770 cm<sup>-1</sup> on the blue wing of the Cd line 326.1 nm, this means that it's lying at 397 cm<sup>-1</sup> on the red wing of the forbidden Cd line 314.1 nm.

The aim of this work, is to find the interatomic potential of the state  ${}^{3}1_{u}$  and also the dipole moment variation D(R) that are responsible for the formation of the band A. To do this, the interatomic potential of the excited  $({}^{3}1_{u})$  state associating with  $(5{}^{3}P_{2} - 5{}^{1}S_{0})$  of Cd transition and its dipole moment D(R) are modified to get the best fit profile of band A with the experimental one measured by (Helmi *et al* 1996).

#### Materials and Methods

*UFC Profile.* The unified Frank Condon (UFC) model developed by Szudy and Baylis (Szudy and Baylis 1975) is a quasimolecular treatment of spectral line profiles of collision molecules (Szudy and Baylis 1975, Sando and Wormhoudt 1973. For the case of not two strongly varying dipole transition moments, the reduced molecular absorption coefficient k(v, T) in the line wing region of the atomic line  $V_{ij}=E_j-E_i/h$  can be expressed due to UFC theory in the form:

$$K(v,T) = \frac{32\pi^{7/2} v}{c g_o} \sum_i \frac{R_i^2 D^2(R_i)}{|\Delta V(R_i)|} \exp\left[\frac{-Vg(R_i)}{kT}\right] [Z_i]^{1/2} L(Z_i)....(1)$$

where  $g_0$  is the molecular statistical weight of the initial ground state  $(X^{10}_g)$  and  $D(R_i)$  is the molecular transition moment. The summation extends over all degenerate substates of the initial  $X^{10}_g$  and final state  ${}^{31}_{u}$ .  $R_i$  denote Condon points, that are complex solutions of:

$$h v = \Delta V(R_i) = V_i(R_i) - Vg(R_i)$$
 .....(2)

The real  $R_i$  represent the transition points, that obey the Frank-Condon principle. Z is the reduced frequency given by:

$$Z_{i} = 0.5 \left(\frac{\mu}{kT}\right) \left[\frac{\Delta V (R)}{h}\right]^{-2} \left|\frac{\Delta V (R)}{h}\right|^{-4/3} \dots (3)$$

 $L(Z_i)$  is the universal line shape function defined and tabulated in (Szudy and Baylis 1975),  $\mu$  is the reduced mass of the collision molecule and T is the temperature.  $\Delta V$  and  $\Delta V^{\circ}$  are the first and second derivatives of the difference potential  $\Delta V(R)$  respectively.

*Procedure of calculation.* It is seen from Eq.(1) that to calculate the molecular band A profile, we need to know the

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ground state potential  $V_g(R)$  for  $Cd_2$  system, the difference potential  $\Delta V(R)$  between the initial and final states and the dipole moment D(R). We have chosen for the ground state potential  $V_g(R)$  of the quasimolecule  $Cd_2$ , the potential determined experimentally by (Helmi *et al* 1996). This potential is then fitted to the Morse potential by appropriate parameters. The function  $\Delta V(R)$  corresponding to the potential difference between  $[V_g \text{ and } {}^31_u]$  states leading to the red wing of the Cd line 314.1 nm profile and also the dipole moment D(R) are chosen that give the best fit between the calculated and the experimental profiles measured by (Helmi *et al* 1996 and Grycuk *et al* 1992). As  $\Delta V(R)$  and  $V_g$  are known, it is easy to obtain the excited state  $(V_1)$  potential. This potential is also fitted to the Morse potential by appropriate parameters.

#### **Results and Discussion**

The experimental line profiles of the bands A, B and C obtained by (Helmi *et al* 1996 and Grycuk *et al* 1992) for  $Cd_2$ interaction at about 770, 830 and 1800 cm<sup>-1</sup> on the blue wing from the centre the Cd (326.1 nm) line are presented in Fig 1.

The analytical form of the function  $\Delta V(R)$  has been chosen so that its extremity ( $\Delta V_m = h\Delta v_m$ ), where  $\Delta v_m = 397 \text{ cm}^{-1}$  is the spectral position of the band A on the red wing of the forbidden Cd (314.1 nm ) line. Figure 2 shows the effect of curvature of the function  $\Delta V(R)$  near the vicinity of its extrumum (A) on the profile of the band A (B). The sensitivity of the profile to the curvature of  $\Delta V(R)$  implies that the shape of the  $\Delta V(R)$  is accurately determined, which is not the case for the effect of D(R) as shown in Fig 3 A & B. This is in agreement with the same effects applied on the Cs satallite published in (Visticot *et al* 1981). Figures 4,5 give respectively the functions D(R) and  $\Delta V(R)$  that give the best fit calculated profile for band A with the observed one. This is illustrated in Fig 6. The ground state potential  $V_g(R)$  for Cd<sub>2</sub> interaction has been taken from the author's papers (Helmi *et al* 1996; Roston and Helmi 1998). The fitting parameter  $\beta$ 

Table 1Comparison of our data for the well depth  $\epsilon$ (cm<sup>-1</sup>) andits position  $R_m$  (A) of the state  ${}^{3}1_u$  with other data

<sub>u</sub> State <sup>3</sup> 1		Reference
R <sub>m</sub> (A)	ε(cm <sup>-1</sup> )	
3.3	844.5	Grycuk et al 1992
3.2	799.5	



Fig 1. Expeprimental blue wing profile of Cd line 326.1 nm (Helmi et al 1996 and Grycuk et al 1992).



Fig 2. The calculated profile of the band A(B) corresponding to the curvature variation function of  $\Delta V(R)$  (A), at the same extremum of  $\Delta V(R)$  and the dipole moment.



Fig 3. The calculated band profile (B) corresponding to the variation of the dipole moment D(R) (A), at the same curvature and extremum of V(R).



**Fig 4.** The obtained induced dipole moments (1) this work, (2) by quantum calculations.



Fig 5. The difference  $\Delta V(R)$  with the excited potentials obtained in this work and the ground  $(V_g)$  state potential taken from (Helmi et al 1996, Roston and Helmi 1998).



**Fig 6**. (1) The experimental band A profile obtained by (Helmi et al 1996 and Grycuk et al 1992) and (2,3) are respectively the calculated profiles by the authors and that obtained by (Grycuk et al 1992 using quantum calculation.

of this potential to the Morse potential with its depth  $\varepsilon_g = 386 \text{ cm}^{-1}$  (Helmi *et al* 1996) and depth position  $R_g = 3.7 \text{ A}$  [12] is obtained as  $\beta = 1.1$ .

As the ground state potential  $V_g(R)$  and  $\Delta V(R)$  are known, then the excited state potential  $V_1$  for the molecular state  ${}^{3}1_u$ of the  $(5{}^{3}P_2 \rightarrow 5{}^{1}S_o)$  transition for Cd<sub>2</sub> system can be obtained using the relation:

$$V_1(R) = V_o(R) + \Delta V(R)$$

The Morse potential parameters of the obtained excited state potential  $V_1(R)$  are found to be equal to  $\varepsilon_1 = 844.5 \text{ cm}^{-1}$ ,  $R_1 = 3.3 \text{ A}$  and  $\beta = 1.3$ . The ground  $V_g(R)$  and the excited  $V_1(R)$  state potentials with the difference potential  $\Delta V(R)$  are presented in Fig 5. Figure 4 shows the induced dipole

moment D(R) for the molecular transition  $(X^{10}_g \rightarrow {}^{3}1_u)$  obtained by the authors that gives the best fit for the calculated and experimental profiles of the band A and dipole moment for that transition obtained by (Grycuk *et al* 1992). The obtained D(R) has an exponential form given by:

$$d(r) = A \exp\left(\frac{\frac{R-R_o}{b}}{b}\right)^2$$

where A, b and R are respectively equal to 1, 0.8, 3.

## Conclusion

The band A appearing at  $\Delta v = 770$  cm<sup>-1</sup> on the blue wing of the Cd (326.1 nm) line broadened by cadmium pressure has been analysed using (UFC) model of Szudy and Baylis (1975). This analysis has led to the following conclusions.

i)The shape of the band A is strongly affected by the curvature of the difference potential  $\Delta V(R)$  in the vicinity of its extrumum, while it is little affected by the dipole moment variation D(R).

ii) The dipole moment variation D(R) obtained by the authors is in close agreement with D(R) obtained before by (Grycuk *et al* 1992) using quantum calculations.

iii) The obtained interatomic potential of the excited state  $({}^{3}I_{u})$  resulting from  $({}^{3}P_{2} - {}^{1}S_{0})$  for Cd<sub>2</sub> interaction using this analysis is in close agreement with the potential of that state obtained by (Grycuk et al 1992) using quantum calculations. This is shown in the Table 1.

iv) From the foregoing discussion it is seen that the analysis of the molecular spectrum using (UFC) model gives reasonable results for the interatomic potential of the molecular excited state  $(^{3}l_{u})$  comparing with other procedures for obtaining such potential.

## References

- Bender C F, Rescigno T N, Schaefer H F, Orel A H 1979 Potential energy curves for diatomic Zinc and Cadmium, *J Chem Phys (USA)* 71(2) 651.
- Bousquet C 1986 Absorption profile of 228.8 nm Cd resonance line broadened by pressure effects of Cd and Kr: Cd-Cd and Cd-Kr interaction potentials. J Phys B At Mol Phys 19 3859-3879.
- Czajkowski M, Bobkowski R, Krause L 1989  $o_u^+({}^3\Pi_u) \leftarrow Xo_g^+(\Sigma_g^+)$  Transition in Cd<sub>2</sub> excited in crossed molecular Laser Beams. *Phys Rev* **40** 4338.
- Grycuk T, Kowalczyk L, Morawski Z 1992 Absorption Spectrometry of the Cd<sub>2</sub> Eximer. SPIE, High-Performance Optical Spectrometry, 1711, 159.
- Helmi M S, Grycuk T, Roston G D 1996 Interatomic Potentials and the Van der Waals Coefficient for the intercombination Cd 326.1 nm absorption line broadened by Cadmium Pressure. J Spec Chem Acta B51 633.
- Kowalski A, Czajkowski M, Breckendrige W H 1985 Laser excitation Spectrum of the  $o_g^+(\Sigma_g^+) \leftarrow o_u^+({}^3\Pi_u)$  transition in Cd<sub>2</sub>. Chem Phys Lett **119** 368-370.
- Roston G D, Helmi M S 1998 Determination of the ground state well depth position Rm of Cd-inert gases molecules experimentally. 14th ICSLS conference, State College Pennsylvania, USA, 462.
- Stevens W J 1979 Cd<sup>\*</sup><sub>2</sub> as a 470 nm absorber. Appl Phys Lett 35 751.
- Szudy J, Baylis W E 1975 Unified Franck-Condon Transition of Pressure Broadening of Spectral Lines. JQSRT 15 641.
- Sando K M, Wormhoudt J C 1973 Semiclassical shape of satellite bands. *Phys Rev* A7 1889.
- Visticot J P, Suzdy J, Sayer B 1981 Semiclassical description of the satellite profile of the Cs  $(6S_{1/2} \rightarrow 5D_{1/2})$  Transition perturbed by Argon allowing molecular Potential Determination. J Phys B At Mol Phys **14** 2329-2335.