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Study of Transport Properties of Mineral Chalcopyrite (CuFeS₂) at Relatively Low Temperatures (77-300 K)

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Abstract. The electrical resistivity and thermoelectric power measurements were carried out on chalcopyrite; the mineral rock sample of Balochistan, Pakistan, in a temperature range down from liquid nitrogen temperature up to room temperature (77-300 K). The resistivity studies show diverse transitions at different temperature intervals especially a semisuperconductivity, where weak fermions (not the Cooper pairs) are produced. This transition is accompanied with another relatively low temperature transition, i.e., a metal-insulator transition which sets in at 92 K. The metal to insulator transition at 92 K occurs due to breakup of weak fermions. The Grüneisen function shows a transition at 215 K, where some kind of phonons become free from their respective Brillouin zones. Thus chalcopyrite at 215 K can produce excited phonons for semiconductor, imaging, nanotube, quantum well, optoelectronic and other devices. The sudden escalation of positive thermoelectric power at 240 K in chalcopyrite shows a phase transition perhaps due to diffusional recovery processes, i.e., shuffling of electrons and positive ion cores. Thermoelectric power measured shows an increasing trend similar to that of the theoretically estimated values with a room temperature value around 600 μ V/K.

Keywords: transport mechanism, mineral ore, chalcopyrite, electrical resistivity, thermoelectric power.

Introduction

The electrical resistivity of liquid quenched amorphous alloys has been previously studied (Naqvi *et al.*, 1995a; 1995b; Raza *et al.*, 1995) in which methods for estimation of Debye temperature, Debye wave vector, Debye velocity, Fermi wave vector, Fermi velocity, Fermi energy, relaxation time, mean free path, temperature coefficient and activation energy were presented. Use of Arrhenius plot for calculations of activation energy has indicated some thermally activated processes in metallic glasses, which would decipher the temperature dependent activation energies accompanying multiple transitions (Naqvi *et al.*, 1991). Metal-insulator transition in iron-zirconium alloys were observed at relatively high temperatures (Naqvi *et al.*, 1997).

Our previous studies tuned us to study the rock minerals, pyrite (FeS₂) (Naqvi *et al.*, 2007) and chalcopyrite (CuFeS₂). Earlier, we had analyzed the ore samples of Balochistan, Pakistan by X-ray diffraction (XRD); magnetic properties of pyrite (FeS₂) and chalcopyrite (CuFeS₂) from these ore samples were also studied in the temperature range of 77-300 K, showing an overall paramagnetic nature of both pyrite and chalcopyrite (Naqvi *et al.*, 2005). The electron density distributions of pyrite (FeS₂) and chalcopyrite (CuFeS₂) and chalcopyrite (CuFeS₂) have been recently studied by Gibbs *et al.* (2007).

several other interesting properties. Kametani (1982a), gave a method for thermoelectric power measurement of pulverized metal sulphide ores, including chalcopyrite. Thermoelectric power of pulverized samples of chalcopyrite from various mines was measured (Kametani, 1982b), indicating its *n*-type semiconducting character; the ionization energy obtained was 0.1-0.3 eV. In another study Kametani and Kobayashi (1990) measured the thermoelectric power for particulate chalcopyrite samples from Mount Isa mine. The thermal expansion of synthetic CuFeS₂ was studied at 4-300 K by Sirota et al. (1989a); at 5-300 K, the coefficients were negative; the Grüneisen coefficient was also determined. Sirota et al. (1991; 1989b) also studied lattice dynamics of chalcopyrite in this low temperature range by X-ray diffractometry, determining the Debye-Waller temperature factor and mean square atomic displacements for the ions. Nkoma and Ekosse (1999) applied X-ray diffraction technique to the ore samples and obtained the lattice parameters for chalcopyrite. The phonon and magnon dispersion of chalcopyrite has been reported by Harris et al. (1997). Its electronic structure has been studied by Kurmaev et al. (1998) and Fujisawa et al. (1994). Donovan and Reichenbaum (1958) compared the electrical properties of synthetic and natural chalcopyrite. Teranishi (1962; 1961) studying the magnetic and electrical properties of chalcopyrite found that it had low electrical conductivity.

Apart from being a major ore of copper, chalcopyrite has

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In a recent study, these properties in the temperature range of 77-300 K were studied by Khabibullin *et al.* (2009) along with NMR spectrum. They found different low temperature conductivity behaviour for different samples. Semiconductors of chalcopyrite have several applications, such as in photo voltaic devices, solar batteries, non-linear optical devices and luminescence diodes (Shay and Wernick, 1975). Chalcopyrite has its applications in most modern devices ranging from imaging devices, quantum nanodevices and optoelectronic devices (Steiner *et al.*, 2009; Ishii *et al.*, 2008; Kohara *et al.*, 2005; Vincent *et al.*, 2005; Lee *et al.*, 2003; Bai *et al.*, 2002).

In the present work, measurements of thermoelectric power and electrical resistivity of chalcopyrite (CuFeS₂), in the temperature range, 77-300 K have been carried out. Grüneisen function (Barnard, 1972; Rosenberg, 1963) and various other parameters have also been calculated (Table 1 and 2) using the technique employed in our previous work on chalcopyrite (CuFeS₂) (Naqvi *et al.*, 1997; Raza *et al.*, 1995; Barnard, 1972).

Materials and Methods

The rock samples of chalcopyrite were cut in the shape of parallelepiped of size $2\times6\times10$ mm. The standard four probe method was employed for the electrical resistivity measurement on polished samples of chalcopyrite. A constant current source, Digistant-6426 and Keithley multimeter, 175 & 197A were used in the low temperature measurement of electrical resistivity. Thermoelectric power measurements were carried out using the differential method, as described elsewhere (Naqvi *et al.*, 1993).

Results and Discussion

Electrical resistivity vs temperature plot of chalcopyrite is shown in Fig. 1. Fig. 2, represents the Arrhenius plot (log p $vs \frac{1000}{T}$) for electrical resistivity of chalcopyrite. Fig. 3 and 4 show Grüneisen function vs temperature and thermoelectric power vs temperature curves, respectively, for chalcopyrite, in the range, 77-300 K. With Arrhenius plot, the transition regions can be studied on the basis of theory of Raza et al. (1995). The labelled transition regions of electrical resistivity at various temperatures are shown in Fig. 1. AB-transition shows metallic behaviour, BB'- semi-metallic, where the conduction electrons are trapped, B'C- metal softening, where the deep energy levels are becoming shallow, CDsemi-conductivity, where the onset of semi-conductivity at 208 K is observed and DE- semi-superconductivity, where weak fermions are produced (not the Cooper pairs). With lowering of temperature from 120 K to 80 K, transition-EF was observed which shows a sudden increase in electrical resistivity thereby indicating that the weak fermions are broken at low temperatures, i.e., at about 92 K. EF is ascribed to a metal-insulator transition. This is evident by mean free path of 102.24 m at 92 K (at E; end of DE region). These transitions can be ascertained with the help of Arrhenius plot as shown in Fig. 2.

The Bloch-Grüneisen function, \mathscr{G} (Barnard, 1972; Rosenberg, 1963) was calculated using:

$$\mathscr{G} = \text{constant} \times \frac{T^5}{M\theta^6} \int_0^{\theta/T} \frac{z^5 \, dz}{(e^z - 1)(1 - e^{-z})}$$
(1)

where:

- T = temperature in Kelvin,
- θ = Debye temperature,

z = atomic number,

- M = atomic weight and
- constant = characteristic of the metal.

A spherical fermi-surface was assumed, neglecting the Umklapp processes. Thus, a Debye-type lattice spectrum is being dealt with. At high temperatures, $(T > \frac{\theta}{2})$, equation (1) get reduced to:

$$\mathscr{G} = \text{constant} \times \frac{\text{T}}{4\text{M}\theta^2}$$
(2)

and at low temperatures (T< θ /10), the upper limit of the integration in equation (1) can be taken as infinity and equation (3) may be obtained:

$$\mathscr{G} = \text{constant} \times \frac{124.4\text{T}^5}{\text{M}\theta^6}$$
(3)

where the constant for each transition was calculated from Fig. 1, i.e., by employing $\frac{\Delta \rho}{\Delta T}$, θ the Debye temperature and M the atomic weight of chalcopyrite (CuFeS₂) i.e., 184 amu. Equation (3) was used for the Fig. 3. The plot of Grüneisen function in Fig. 3 is very interesting. It shows a sharp transition at 215 K, which is indicative of the fact that some kind of phonons, may be optical or acoustical, are excited at 215 K and become free from their respective Brillouin zones.

Figure 4 shows both theoretical and experimental results of thermoelectric power of chalcopyrite. The thermoelectric power undulations in chalcopyrite are increasing with increasing temperature. There is some kind of phase transition occurring at 240 K, where the maximum positive thermoelectric power is observed in the temperature range of 77-300 K. The theoretical thermoelectric power, S_{th}, is also calculated which shows a monotonic linear trend. The chalcopyrite sample, from Balochistan, has a room temperature thermopower value of 600 μ V/K; as compared to -260

Table 1. T chalcopyrit	he calcula e	ted values of activ	ation ener	rgy, relaxation time, me	an free path and	Grüneisen functio	n at all labelled t	ransition temperat	ure regions for
Temperature T with labelled	Δ Δ T	Grüneisen function Ø=	$\frac{\Delta \log \rho}{\Delta (10^3/T)}$	Activation energy (from Arrhenius plot) with Boltzmann constant k.=8.62x10 ³ eV/K	Activation energy (from Berry)	Temperature coefficient	Relaxation time (from plot) witht $t=7.1 \times 10^{-4}$ s	Relaxation time (from Berry)	Mean free path with $v=v_D$ =3.41×10 ⁴
transition regions		$\frac{\left(\frac{\Delta\rho}{\Delta T}\right)\!124.4T^{5}}{M\theta^{6}}$	×10 ⁻³	$Q_{o} = \frac{\left(\frac{\Delta \log p}{\Delta(10^{3}/\mathrm{T})}\right) 10^{3} k_{B}}{\log e}$	$Q(T){=}Q_{o}(1{-}\alpha T)$	with $t_o = 10^{-13} s$ $\alpha = -\left(\frac{k_B}{Q_o}\right) log\left(\frac{t_o}{t_o}\right)$	$\mathbf{t}_{0}^{'} = \frac{\mathbf{t}}{\exp\left(\frac{\mathbf{Q}_{0}}{\mathbf{k}_{\mathrm{B}}T}\right)}$	$t=t_{o}^{'}exp\left(\frac{Q_{o}}{k_{B}T}\right)$	m/s^{-1} 1 = wt_o^{-1}
(K)	$\times 10^{-4} \\ (\Omega \text{-m/k})$	(Arbitrary units)	(eV)	(eV)	(K ⁻¹)	(s)	(s)	(m)	
92 E-F	138.54	3.1E-11	941.2	0.186846	0.183824	0.000176	4.16E-14	0.00071	5.989E-09
183 D-E	ı				ı	ı	0.00071	0.00071	102.24
209 C-D	21.6	2.93E-10	1599.7	0.31757	0.35713	-0.0006	1.57E-11	0.00071	2.260E-06
215 B`-C	-93.6	1.46E-09	8144.9	1.616914	1.097271	0.001495	9.14E-42	0.00071	1.317E-36
221 B-B`	34.48	6.17E-10	5628.8	1.117421	0.8198	0.001205	2.38E-29	0.00071	3.430E-24
300 A-B	1.13	9.33E-11	108.73	0.021585	0.266964	-0.03789	0.000308	0.00071	44.37334
Table 2. Tł	ie calculate	ed values of Debye	temperatuı	re, Debye wave vector, D	ebye velocity, Fe	mi wave vector, Fe	rmi velocity and H	Fermi energy for ch	lalcopyrite
Sample	Dei	bye	Debye	Debye		Fermi	Fen	mi	Fermi
	wa	ve vector	velocity	temperatur	e	wave vector	velo	ocity	energy
		h = $4.25 \times 10^{26} \text{ m}^{-3}$	with $\omega_{\rm c} = 10^1$	with 14 rad/s 14 16 16 16 17 14	< 10 ⁻³⁴ Is	with $\overline{N} = 4.25 \times 10^{-10}$	with 1 ²⁶ m-3 ħ =	h 1 0552 × 10 ⁻³⁴ Is	
	>		OT Com	kB = 1.38 >	$\times 10^{-23}$ J/K		m	1.0022×10^{-31} kg	

Sample	Debye	Debye	Debye	Fermi	Fermi	Fermi
	wave vector with $\frac{N}{V} = 4.25 \times 10^{26} \text{ m}^3$	velocity with $\omega_{\rm b} = 10^{14}$ rad/s	temperature with $h = 1.055 \times 10^{-23}$ JK kB = 1.38 × 10^{-23}JK	wave vector with $\frac{N}{V} = 4.25 \times 10^{26} \text{ m}^{-3}$	velocity with $h = 1.0552 \times 10^{34}$ Js $m = 9.11 \times 10^{31}$ kg	energy
	$k_D = \left(6\pi^2 \frac{N}{V}\right)^{1/3}$	$v_{\rm D} = \frac{\omega_{\rm D}}{k_{\rm D}}$	$\Theta_{\text{theo}} = \left(\frac{\hbar v_{\text{D}}}{k_{\text{B}}}\right) \left(6\pi^2 \frac{N}{V}\right)^{1/3}$	$k_{\rm F} = \left(3\pi^2 \frac{N}{V}\right)^{1/3}$	$v_{\rm F} = \frac{\hbar k_{\rm F}}{{ m m}}$	$E_{\rm F} = \frac{\hbar^2 k_{\rm F}^2}{2m}$
	(m ⁻¹)	(m/s)	(K)	(m ⁻¹)	(m/s)	(eV)
Chalcopyrite (CuFeS ₂)	6.94×10 ⁸	1.44×10 ⁵	764.14	5.51×10 ⁸	6.38×10 ⁴	0.0115

Transport Properties of Mineral Chalcopyrite

119

and -290 μ V/K; obtained by Kametani *et al.* (1990) from the chalcopyrite samples of Mount Isa mine and 120 μ V/K by Izmailov and Silantev (1978). The higher value of thermoelectric power obtained might be due to pyrrhotite (Fe_{1-x}S (x = 0 to 0.2)), which plays a significant role in the conductivity of chalcopyrite as observed by Izmailov and Silantev (1978).



Fig. 1. Electrical resistivity *vs* temperature for chalcopyrite (CuFeS₂).



Fig. 2. Arrhenius plot for activation energies of chalcopyrite (CuFeS₂) i.e., log p vs 1000/T.



Fig. 3. Grüneisen function *vs* temperature for chalcopyrite (CuFeS₂).



Fig. 4. Experimental and theoretical curves showing values of thermoelectric power (S) *vs* temperature for chalcopyrite (CuFeS₂).

There are many cases for estimating the theoretical thermoelectric power (Barnard, 1972), but we shall restrict to diffusion thermopower for a free electron metal or alloy obeying Fermi-Dirac statistics in which $E_F >> k_B T$ (the degenerate case) and in which a single relaxation time exists. The theoretical thermoelectric power for a range of representative value of E_F is given by the following equation.

$$S_{th} = -\frac{\pi^2 k_B^2 T}{3|e|E_F}$$
(4)

where:

 $k_B = Boltzmann constant and$

e = charge of an electron.

Equation (4) is valid only if mean free path is regarded as a constant. The model of constant mean free path is applicable to electron scattering by impurity atoms and equation (4) is reduced to,

$$S_{th} = -\frac{2.45 \times 10^{-8} T}{E_{F}}$$
(5)

where:

 E_{F} is in electron volts and T is in Kelvin.

Table 1 and 2 represent the calculated values and the formulae used for the parameters. The techniques used for such calculations, as mentioned in Table 1 and 2, are taken from the previous studies (Naqvi *et al.*, 1997; Raza *et al.*, 1995), excepting the Grüneisen function (Barnard, 1972; Rosenberg, 1963). Sulphur, being a constituent element of chalcopyrite, has two allotropes known as rhombic and monoclinic sulphur. Rhombic sulphur is energetically stable below 369 K. Hence, the sulphur in chalcopyrite exists in the form of rhombic allotrope, in the temperature range, 78-300 K for electrical resistivity and thermoelectric power.

Conclusion

Summarizing the results, diverse transitions at various temperatures are observed in the study of electrical resistivity and thermoelectric power of chalcopyrite from room temperature down to liquid nitrogen temperature.

A semi-superconductivity transition in chalcopyrite is associated with formation of weak Fermions (not the Cooper pairs). A metal-insulator transition sets in at 92 K due to breakup of weak Fermions. Some kind of phonons, may be optic or acoustic, are excited at 215 K in chalcopyrite. Various semiconductor, imaging, nanotube, quantum well, optoelectronic and other devices based on this phenomenon can be designed using chalcopyrite (Steiner *et al.*, 2009; Ishii *et al.*, 2008, Kohara *et al.*, 2005; Vincent *et al.*, 2005, Lee *et al.*, 2003; Bai *et al.*, 2002). A phase transition due to diffusional recovery occurs at 240 K, indicated by the thermo-electric power study. A significantly high value of around 600 μ V/K, room temperature thermopower was observed in the studied chalcopyrite sample.

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